

STEREOCHEMISTRY OF IODOFLUORINATION OF PHENYLSUBSTITUTED  
OLEFINS WITH METHYLIODINE(III)DIFLUORIDE

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Recently, we have found that methyl iodine(III) difluoride reacts with 1,1-diphenylethylenes in the presence of hydrogen fluoride as catalyst to form corresponding 1-fluoro-2-iodo-1,1-diphenylethanes<sup>1</sup>. We now report the results of a similar addition to *cis*- and *trans*-1-phenylpropene-1 and *cis*- and *trans*-stilbene with this reagent. The recently reported synthesis of methyl iodine(III)-difluoride<sup>2</sup> by the reaction of xenon difluoride with an excess of methyl iodine made this potential fluorinating agent readily available. The attempted fluorination of *trans*-stilbene with aryl iodine(III) difluorides<sup>3</sup> failed, but reaction with *trans*-1-phenylpropene-1 resulted in the rearranged product 1,1-difluoro-2-phenylpropane<sup>4</sup>.

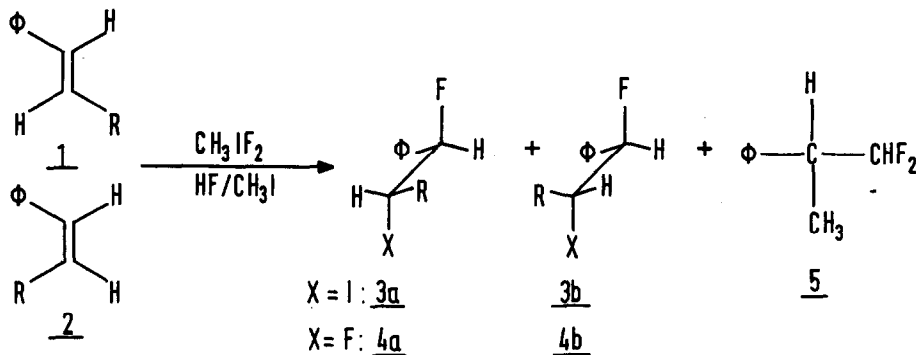
Being interested in learning whether each member of a *cis* and *trans* isomeric olefin pair would exhibit the same stereochemistry in the iodine fluoride addition with methyl iodine(III) difluoride, we chose *trans*-(1) and *cis*-stilbene(2) (R = Phenyl) and 1-phenylpropene-1 (R = CH<sub>3</sub>) for the study. In a typical experiment, anhydrous HF was introduced at 20<sup>o</sup> in 2 ml methyl iodide and under stirring pure XeF<sub>2</sub> (1.2 mmol) was added. The solution turned dark brown and xenon gas was evolved. After 15 minutes, when gas evolution had ceased, olefin (1 mmol) was added. Two hours later the products were isolated, <sup>1</sup>H and <sup>19</sup>F n.m.r. spectra were taken on the crude reaction mixtures, and the products (3a), (3b), (4a), (4b) and (5) were separated by g.l.c. or t.l.c. The conversions of 1 and 2 into products were of the order of 90%, as determined by n.m.r. or g.l.c.

The reaction of methyl iodine(III) difluoride with *cis*- and *trans* stilbene gave a reaction mixture of diastereoisomeric iodofluorides (3a) and (3b) and difluorides (4a) and (4b). The product distribution data indicate that the iodofluorination of both *cis*- and *trans*- stilbene is a rather highly *anti* stereoselective process. The formation of difluorides exhibits lower *anti* stereoselectivity. The reaction with *cis*- and *trans*-1-phenylpropene-1 gave a reaction mixture of diastereoisomeric iodofluorides (3a) and (3b), trace amounts of difluorides (4a) and (4b) and about 15% of rearranged geminal difluoride (5). As in the iodofluorination of stilbenes, a high *anti* stereoselectivity and exclusive Markovnikov type regioselectivity of the addition of iodine fluoride was observed.

The structures of the previously known products (4a) and (4b) were identified<sup>5,6</sup> and those of unknowns (3a) and (3b) assigned by their i.r., <sup>1</sup>H and <sup>19</sup>F n.m.r. and mass spectra, and the stereospecific anti elimination of HI affording known *cis* and *trans* phenylsubstituted fluoroolefines<sup>6</sup>.

Isomerisation of the reacting olefins was evaluated in separate experiments stopped prior to completion. The unreacted olefin was analysed by g.l.c. Isomerisation of olefins 1 and 2 during the course of reaction is negligible. The amount of difluorides 4a and 4b or 5 is greater if a longer reaction time is used. Difluorides were formed only from olefins bearing at least one methyl or phenyl group (R)<sup>1</sup>, which facilitates the S<sub>N</sub>1 substitution with its mesomeric or inductive effect. Following some reactions with <sup>19</sup>F n.m.r., we observed first the formation of iodofluorides only, but in the later stages of the reactions <sup>19</sup>F peaks corresponding to vicinal or rearranged geminal difluorides appeared.

Methyl iodine(III) difluoride appears to be a mild, selective agent for the iodofluorination of olefins, with some advantages in comparison to known reagents e.g. anhydrous hydrogen fluoride<sup>7</sup> or Olah's<sup>8</sup> poly-hydrogen fluoride/pyridine in connection with N-iodosuccinimide. One of them which could be mentioned here is the only catalytic amount of HF required to catalyse the reactions, which prevents the polymerisation of rather sensitive olefins or other side reactions.

RELATIVE YIELDS (%)<sup>a</sup>

	R	3a	3b	4a	4b	5
<u>1</u>	CH <sub>3</sub>	78	4	4	—	14
	Φ	56	4	36	3	—
<u>2</u>	CH <sub>3</sub>	4	80	—	1	15
	Φ	10	52	20	18	—

<sup>a</sup>Determined by <sup>19</sup>F n.m.r. spectroscopy

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## REFERENCES

- 1) M.Zupan and A.Pollak, in press.
- 2) J.A.Gibson and A.F.Janzen, J.Chem.Soc., Chem.Comm., 1973, 739.
- 3) W.Bockenmuller, Chem.Ber., 1931, 64, 522.
- 4) M.Zupan and A.Pollak, unpublished observations.
- 5) M.Zupan and A.Pollak, Tetrahedron Lett. 1974, 1015.
- 6) R.F.Merritt, J.Amer.Chem.Soc., 1967, 89, 609.
- 7) A.Bowers, J.Amer.Chem.Soc., 1959, 81, 4107.
- 8) G.A.Olah, M.Nojima and I.Kerekes, Synthesis, 1973, 780.

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